

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
11 April 2002 (11.04.2002)

PCT

(10) International Publication Number
WO 02/28548 A2

(51) International Patent Classification⁷: **B05D**

Jas, Pal [GB/GB]; Redgate House, Wolsingham, County Durham DH1 3LE (GB). **WARD, Luke** [GB/GB]; 80 Hallgarth Street, Durham DH1 3AY (GB).

(21) International Application Number: PCT/GB01/04272

(22) International Filing Date:
25 September 2001 (25.09.2001)

(74) Agents: **DONLAN, Andrew, M.** et al.; Dow Corning Limited, Cardiff Road, Barry CF63 2YL (GB).

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
0024230.5 4 October 2000 (04.10.2000) GB
0114877.4 19 June 2001 (19.06.2001) GB

(81) Designated States (*national*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW.

(71) Applicant (*for all designated States except US*): **DOW CORNING CORPORATION** [US/US]; Midland, MI 48611 (US).

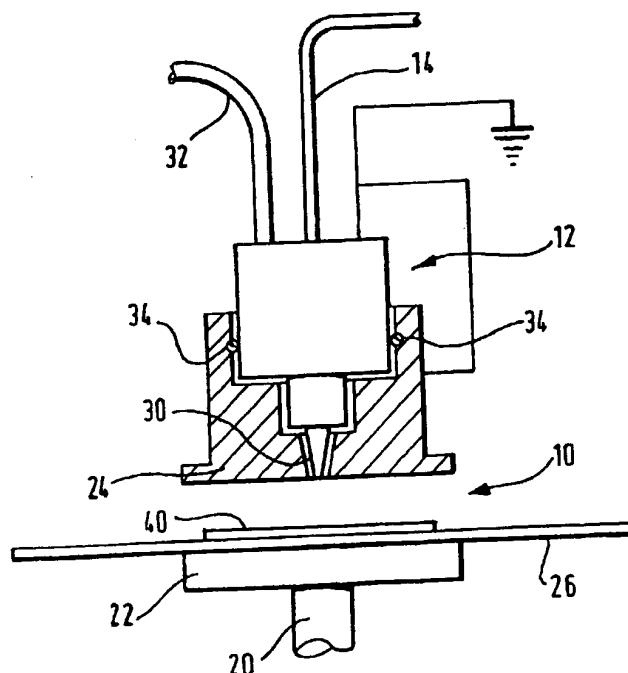
(84) Designated States (*regional*): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

(72) Inventors; and

(75) Inventors/Applicants (*for US only*): **GOODWIN, Andrew** [GB/GB]; 16 Grangewood Close, Pontprennau, Cardiff CF23 8PP (GB). **MERLIN, Patrick** [BE/BE]; Rue Hubermont 10, B-7063 Neufvilles (BE). **BADYAL,**

[Continued on next page]

(54) Title: METHOD AND APPARATUS FOR FORMING A COATING



(57) Abstract: A method for forming a coating on a substrate using an atmospheric pressure plasma discharge. The method comprises introducing an atomised liquid and/or solid coating-forming material into an atmospheric pressure plasma discharge and/or an ionised gas stream resulting therefrom, and exposing the substrate to the atomised coating-forming material. The application also describes a method for polymerising a polymer forming material, and further to apparatus for forming a coating on a substrate.

WO 02/28548 A2

**Declarations under Rule 4.17:**

- as to applicant's entitlement to apply for and be granted a patent (Rule 4.17(ii)) for the following designations AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE,

DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG)

- of inventorship (Rule 4.17(iv)) for US only

Published:

- without international search report and to be republished upon receipt of that report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

METHOD AND APPARATUS FOR FORMING A COATING

[0001] The present invention relates to a method for forming a coating on a substrate, in particular a method for forming a coating on a substrate using an atmospheric pressure plasma discharge, a method for polymerising a polymer forming material, and further to apparatus for forming a coating on a substrate.

[0002] Substrates may be coated for a variety of reasons, for example to protect the substrate from corrosion, to provide a barrier to oxidation, to improve adhesion with other materials, to increase surface activity, and for reasons of biomedical compatibility of the substrate. A commonly used method for modifying or coating the surface of a substrate is to place the substrate within a reactor vessel and subject it to a plasma discharge. Many examples of such treatment are known in the art; for example, US patent number 5,876,753 discloses a process for attaching target materials to a solid surface which process includes affixing carbonaceous compounds to a surface by low power variable duty cycle pulsed plasma deposition, and EP-A-0896035 discloses a device having a substrate and a coating, wherein the coating is applied to the substrate by plasma polymerisation of a gas comprising at least one organic compound or monomer. DE 19924108, which was first published after the initial priority date of the present application, describes a process for coating dyestuffs and corrosion inhibitors onto substrates. The process involves the application of a liquid film coating onto a substrate and a subsequent plasma polymer protective coating. The plasma polymer coating is formed using gaseous monomers and low pressure plasma.

- [0003] However, such plasma surface treatments require the substrate to be under conditions of reduced pressure, and hence require a vacuum chamber. Typical coating-forming gas pressures are in the range 5 to 25 Nm^{-2} (cf. 1 atmosphere = $1.01 \times 10^5 \text{ Nm}^{-2}$). As a result of the requirement for reduced pressure, surface treatments are expensive, are limited to batch treatments, and the coating-forming materials must be in gaseous and/or vapour form in order to maintain conditions of reduced pressure.
- 10 [0004] The present inventors have found that the abovementioned disadvantages of substrate surface plasma treatment can be overcome using a combination of an atmospheric pressure plasma discharge and an atomised liquid and/or solid coating forming material.
- 15 [0005] Thus, according to the present invention there is provided a method for forming a coating on a substrate, which method comprises introducing an atomised liquid and/or solid coating-forming material into an atmospheric pressure plasma discharge and/or an ionised gas stream resulting therefrom, and exposing the substrate to the atomised coating-forming material.
- 20 [0006] It is to be understood that the coating forming material in accordance with the present invention is a material which can be used to make any appropriate coating, including, for example, a material which can be used to grow a film or to chemically modify an existing surface.
- 25 [0007] The present invention also provides a method for polymerising a polymer-forming material, which method comprises atomising the polymer-forming material, and exposing the polymer-forming material to an atmospheric pressure plasma discharge.
- 30

[0008] The present invention further provides apparatus for forming a coating on a substrate, which apparatus comprises means for generating an atmospheric pressure plasma discharge within which, in use, the substrate is placed, an atomiser for providing an atomised coating-forming material within the plasma discharge, and means for supplying a coating forming material to the atomiser.

[0009] Any conventional means for generating an atmospheric pressure plasma glow discharge may be used in the present invention, for example atmospheric pressure plasma jet, atmospheric pressure microwave glow discharge and atmospheric pressure glow discharge. Typically such means will employ a helium diluents and a high frequency (e.g. > 1kHz) power supply to generate a homogeneous glow discharge at atmospheric pressure via a Penning ionisation mechanism, (see for example, Kanazawa et al, J.Phys. D: Appl. Phys. 1988, 21, 838, Okazaki et al, Proc. Jpn. Symp. Plasma Chem. 1989, 2, 95, Kanazawa et al, Nuclear Instruments and Methods in Physical Research 1989, B37/38, 842, and Yokoyama et al., J. Phys. D: Appl. Phys. 1990, 23, 374).

[0010] The coating-forming material may be atomised using any conventional means, for example an ultrasonic nozzle. The atomiser preferably produces a coating-forming material drop size of from 10 to 100µm, more preferably from 10 to 50µm. Suitable atomisers for use in the present invention are ultrasonic nozzles from Sono-Tek Corporation, Milton, New York, USA. The apparatus of the present invention may include a plurality of atomisers, which may be of particular utility, for example, where the apparatus is to be used to form a copolymer coating on a substrate from two different coating-forming materials, where the monomers are immiscible

or are in different phases, e.g. the first is a solid and the second is gaseous or liquid.

[0011] The present invention may be used to form many different types of substrate coatings. The type of coating which is formed on the substrate is determined by the coating-forming material(s) used, and the present method may be used to (co)polymerise coating-forming monomer material(s) onto the substrate surface. The coating-forming material may be organic or inorganic, solid, liquid or gaseous, or mixtures thereof. Suitable organic coating-forming materials include carboxylates, methacrylates, acrylates, styrenes, methacrylonitriles, alkenes and dienes, for example methyl methacrylate, ethyl methacrylate, propyl methacrylate, butyl methacrylate, and other alkyl methacrylates, and the corresponding acrylates, including organofunctional methacrylates and acrylates, including glycidyl methacrylate, trimethoxysilyl propyl methacrylate, allyl methacrylate, hydroxyethyl methacrylate, hydroxypropyl methacrylate, dialkylaminoalkyl methacrylates, and fluoroalkyl (meth)acrylates, methacrylic acid, acrylic acid, fumaric acid and esters, itaconic acid (and esters), maleic anhydride, styrene, α -methylstyrene, halogenated alkenes, for example, vinyl halides, such as vinyl chlorides and vinyl fluorides, and fluorinated alkenes, for example perfluoroalkenes, acrylonitrile, methacrylonitrile, ethylene, propylene, allyl amine, vinylidene halides, butadienes, acrylamide, such as N-isopropylacrylamide, methacrylamide, epoxy compounds, for example glycidoxypropyltrimethoxysilane, glycidol, styrene oxide, butadiene monoxide, ethyleneglycol diglycidylether, glycidyl methacrylate, bisphenol A diglycidylether (and its oligomers), vinylcyclohexene oxide, conducting polymers such

as pyrrole and thiophene and their derivatives, and phosphorus-containing compounds, for example dimethylallylphosphonate. Suitable inorganic coating-forming materials include metals and metal oxides, including colloidal metals. Organometallic compounds may also be suitable coating-forming materials, including metal alkoxides such as titanates, tin alkoxides, zirconates and alkoxides of germanium and erbium. However, the present inventors have found that the present invention has particular utility in providing substrates with silica- or siloxane-based coatings using coating-forming compositions comprising silicon-containing materials. Suitable silicon-containing materials for use in the method of the present invention include silanes (for example, silane, alkylhalosilanes, alkoxysilanes) and linear (for example, polydimethylsiloxane) and cyclic siloxanes (for example, octamethylcyclotetrasiloxane), including organo-functional linear and cyclic siloxanes (for example, Si-H containing, halo-functional, and haloalkyl-functional linear and cyclic siloxanes, e.g. tetramethylcyclotetrasiloxane and tri(nonofluorobutyl)trimethylcyclotrisiloxane). A mixture of different silicon-containing materials may be used, for example to tailor the physical properties of the substrate coating for a specified need (e.g. thermal properties, optical properties, such as refractive index, and viscoelastic properties).

[0012] In addition, under oxidising conditions the present method may be used to form an oxygen containing coating on the substrate. For example, silica-based coatings can be formed on the substrate surface from atomised silicon-containing coating-forming materials. Under reducing conditions, the present method may be used to form

oxygen free coatings, for example, silicon carbide based coatings may be formed from atomised silicon containing coating forming materials.

[0013] Plasma generating conditions containing gases other than oxygen may also be employed, for example noble gases, air, hydrogen, nitrogen and ammonia. In a nitrogen containing atmosphere nitrogen can bind to the substrate surface, and in an atmosphere containing both nitrogen and oxygen, nitrates can bind to and/or form on the substrate surface. Such gases may also be used to pre-treat the substrate surface prior to exposure to the coating forming substance. For example oxygen containing plasma treatment of the substrate may provide improved adhesion with the applied coating. The oxygen containing plasma being generated by introducing oxygen containing materials to the plasma such as oxygen gas or water. Furthermore, the coating formed on the substrate may be post treated in a range of plasma conditions. For example, siloxane derived coatings may be further oxidised by oxygen containing plasma treatment. The oxygen containing plasma being generated by introducing oxygen containing materials to the plasma such as oxygen gas or water.

[0014] An advantage of the present invention over the prior art is that both liquid and solid atomised coating-forming materials may be used to form substrate coatings, due to the method of the present invention taking place under conditions of atmospheric pressure. Furthermore the coating-forming materials can be introduced into the plasma discharge or resulting stream in the absence of a carrier gas, i.e. they can be introduced directly by, for example, direct injection, whereby the coating forming materials are injected directly into the plasma.

[0015] As mentioned above, the present inventors have found particular utility of the present invention for forming silica- and siloxane-based coatings on substrates using silicon-containing materials. Under oxidising conditions, e.g. an oxygen containing atmosphere, silica-based coatings can be formed on the substrate surface from atomised silicon-containing materials, whereas under non-oxidising conditions a siloxane polymer, e.g. a linear, branched or resinous siloxane polymer, can be formed on the substrate surface from atomisation of a silicon-containing monomer. A siloxane-organic copolymer can be formed on the substrate surface by use of a mixture of organic and silicon-containing monomers. Furthermore, a silica-based coating may be formed on a substrate surface, which may in turn be coated by a further material, for example an organic or siloxane polymer. For example, when a siloxane is mixed with an organic polymer and a substrate formed from said mixture, the siloxane will migrate to the surface of the organic polymeric body of the substrate, due to the difference in surface energy between organic polymers and siloxanes. If this substrate is then subjected to atmospheric pressure plasma treatment, the siloxane on the surface of the substrate is oxidised to form a silica-based coating. This silica-based coating may then be subjected to treatment according to the present invention, by further subjecting it to atmospheric pressure plasma treatment in the presence of atomised silicon-containing monomers, to form a siloxane coating thereon. However, the present invention is also useful for forming an organic coating on a substrate, for example a polyacrylic acid or perfluoro-organic coating.

[0016] The substrate to be coated may comprise any material, for example metal, ceramic, plastics, siloxane, woven or non-woven fibres, natural fibres, synthetic fibres cellulosic material and powder. However, the size of the substrate is limited by the dimensions of the volume within which the atmospheric pressure plasma discharge is generated, i.e. the distance between the electrodes of the means for generating the plasma. For typical plasma generating apparatus, the plasma is generated within a gap of from 5 to 50mm, for example 12 to 25mm. Thus, the present invention has particular utility for coating films, fibres and powders.

[0017] Substrates coated by the method of the present invention may have various utilities. For example, a silica-based coating, generated in an oxidising atmosphere, may enhance the barrier and/or diffusion properties of the substrate, and may enhance the ability of additional materials to adhere to the substrate surface; a halo-functional organic or siloxane coating (e.g. perfluoroalkenes) may increase hydrophobicity, oleophobicity, fuel and soil resistance, and/or the release properties of the substrate; a polydimethylsiloxane coating may enhance water resistance and release properties of the substrate, and may enhance the softness of fabrics to touch; a polyacrylic acid polymeric coating may be used as an adhesive layer to promote adhesion to substrate surface or as part of laminated structure; the inclusion of colloidal metal species in the coatings may provide surface conductivity to the substrate, or enhance its optical properties. Polythiophene and polypyrrole give electrically conductive polymeric coatings which may also provide corrosion resistance on metallic substrates.

[0018] One major problem which tends to occur when coating substrates using a process involving plasma treatment is that the chemical properties of the material used to form the coating may be lost. It is therefore a major advantage of the present invention that the chemical properties of the coating forming material are substantially retained in the coating formed. For example, in the case where acrylic acid is used as the coating forming material, the carboxylic acid functionality is substantially maintained in the coating formed.

[0019] The present invention also provides a method of producing a substrate having a multi-layered coating by the above described processes. In this case a layer of the coating is applied upon each repeat pass of the substrate through the atmospheric plasma glow discharge. Preferably in such a case the substrate may be coated on a continuous basis by being transported through an atmospheric plasma glow discharge by way of a reel to reel process in which the substrate travels from a first reel, through the glow discharge and on to a second reel at a constant speed to ensure that all the substrate has a predetermined residence time within the glow discharge. Each substrate may be subjected to one or more passes through the glow discharge whereby the first or supply reel in the first pass becomes the substrate collecting reel in the second pass and the substrate collecting reel of the first pass in turn is the supply reel in the second pass, the two reels changing over at the end of each pass. Alternatively the substrate may be passed through a series of atmospheric glow discharge chambers.

[0020] Preferred uses of the coatings of the substrates coated in accordance with the present invention include lamination adhesives, oxygen and/or moisture barrier for example for food packaging applications and as a component
5 in or on organic light emitting diode devices in, for example, flat panel displays.

[0021] The present invention will now be illustrated in detail with reference to the accompanying drawing, in which
10 Figure 1 shows an embodiment of apparatus according to the present invention.

[0022] The apparatus according to the present invention shown in Figure 1 comprises means for generating an atmospheric pressure plasma discharge (generally designated
15 10), and an atomiser (generally designated 12) connected to a syringe pump 14 for supplying a coating forming material to the atomiser 12. The means for generating the discharge 10 includes a high voltage 15 kHz ac power supply 20, supplied across two aluminium electrodes 22 and 24 spaced
20 12mm apart, with the lower live electrode 22 shielded by a glass dielectric plate 26. The atomiser 12 includes a Sono-tek* 8700-120 ultrasonic nozzle 30, and is connected to a Sono-tek* 06-05108 broadband ultrasonic generator 32. The atomiser 12 is seated within the earth electrode 24 on an O-
25 ring 34. The substrate 40 to be coated is placed on the glass dielectric plate 26 between the electrodes 22 and 24.

[0023] The apparatus described hereinabove with reference to Figure 1 was used for all the procedures described hereinafter.

30 (* Sono-tek Corporation, Milton, New York 12547, USA).

Example 1

[0024] A piece of polyethylene film substrate was ultrasonically washed in a 1:1 mixture of isopropyl alcohol and cyclohexane and was placed on the glass plate. After evacuation of residual gas, the plasma discharge gas was introduced at a flow rate of 1900 sccm and a pressure of $1.02 \times 10^5 \text{ Nm}^{-2}$. Two discharge gasses were used, helium and a 99% helium/1% oxygen mixture. After 10 minutes of purging, the syringe pump 14 was switched on and the coating-forming material was allowed to flow at a rate of $3 \times 10^{-5} \text{ mls}^{-1}$. Two coating-forming materials were used, octamethylcyclotetrasiloxane (hereinafter referred to as "D₄") and tetramethylcyclotetrasiloxane (hereinafter referred to as "D₄H"). When the coating-forming material reached the ultrasonic nozzle, the ultrasonic generator was switched on (2.5 W) to initiate atomisation of the coating-forming material, and the atmospheric pressure plasma discharge was ignited by applying 1.5 kV across the electrodes. Deposition of the coating-forming material was allowed to proceed for 10 minutes, following which the substrate was removed and placed under vacuum for 20 minutes to remove any labile material.

[0025] The results of the above procedure are shown in Table 1 below. X-ray photoelectron spectroscopic analysis (Kratos ES300) was used to perform elemental analysis of the substrate surface, and a spectrophotometer (Aquila Instruments nkd-6000) was used to determine film thickness. Contact angle measurements were made using video capture apparatus (AST Products VCA2500XE) using sessile $2 \mu\text{l}$ droplets of deionised water.

[0026] Gas permeation measurements of the substrate surface were also taken using a mass spectrometer, and the results are shown in Table 2. The Barrier Improvement Factor is calculated as [coated substrate gas permeation]/[reference sample gas permeation].

Table 1							
Sample	XPS analysis				Contact Angle (°)	Deposition rate (nms ⁻¹)	Coating thickness (nm)
	%C	%O	%Si	%SiO _x			
D ₄ theory	50	25	25	0	-	-	-
D ₄ 100% He	43.3	29.3	25.8		107.8*	28	279
D ₄ 1% O ₂	25.5	48.5	26.0	74.4	56.4	29	286
D ₄ H theory	33.3	33.3	33.3	0	-	-	-
D ₄ H 100%He	32.5	39.1	28.4		102.3	82	
D ₄ H 1% O ₂	9.2	61.4	29.5	81.5	wets	244	

* clean polyethylene has a contact angle of 105.8°

Table 2	
Sample	Barrier Improvement Factor
Clean polyethylene	1.0 (by definition)
D ₄ , 100% He	0.9
D ₄ , 1% O ₂	6.8
D ₄ H, 100% He	0.9
D ₄ H, .1% O ₂	4.5

[0027] ATR-FTIR studies of the substrate surfaces showed that ring-opening polymerisation of the D₄ and D₄H coating-forming materials had occurred to form a polysiloxane on the substrate surface. In particular, the ATR-FTIR studies on the latter showed that the polysiloxane coating retained much of the D₄H Si-H functionality.

[0028] NMR studies of a coating prepared as described above on a glass surface showed that the polysiloxane formed on the substrate surface by polymerisation of the D₄ and D₄H coating-forming materials comprised divalent (CH₃)₂SiO_{2/2} units and trivalent CH₃SiO_{3/2} units, i.e. the polysiloxane is resinous.

15 Example 2

[0029] The method of Example 1 above was repeated using a glass substrate and acrylic acid as the coating-forming material, and helium alone as the discharge gas. The coating was removed from the substrate prior to analysis.

[0030] FTIR and solid state NMR analysis of the coating confirmed that the acrylic acid had polymerised to form polyacrylic acid. Both FTIR and NMR data showed consumption of the unsaturated C=C bond.

Example 3

[0031] The method of Example 2 was repeated, but using nylon and polyethylene substrates.

- 5 [0032] An FTIR analysis comparison of the coating with commercially available polyacrylic acid confirmed that the acrylic acid coating-forming material had polymerised to form a polyacrylic acid coating on the substrate surfaces.

- 10 [0033] X-ray photoelectron spectroscopic analysis, film thickness analysis, and contact angle measurements were performed per Example 1 above. The results are shown in Table 3 below.

Table 3					
	XPS analysis			Contact Angle (°)	Deposition rate (nms ⁻¹)
	%C	%O	%CO ₂ H		
Theory	60.0	40.0	33.3	-	-
Commercial polyacrylic acid	63.3	36.7	29.9	wets	-
Example 3 coating	62.6	37.4	26.4	wets	231 ₊₉₅

- 15 [0034] Gas transport through the coated polyethylene film was determined by mass spectrometry, and the barrier improvement factor calculated per Example 1 above over an untreated polyethylene substrate and commercially available polyacrylic acid. The results are shown in Table 4 below.

Table 4	
Sample	Barrier Improvement Factor
Untreated substrate	1.0 (by definition)
Commercial polyacrylic acid	1.1 \pm 0.1
Example 3 coating	7.2 \pm 0.9

[0035] A lap shear test was performed on the coated nylon substrates as follows. Two opposing faces of coated nylon substrates were overlapped to create a joint covering 1cm², and the substrates were cured under a 2kg weight at 70°C for 60 minutes. The adhesive strength of each joint was then determined by pulling the substrates apart at a rate of 5mm per minute using a tensilometer (Instron), and recording the maximum load reached prior to failure. The coated substrates withstood a maximum load of 74 \pm 11 Ncm⁻² prior to failure. Comparison joints made from uncoated nylon displayed no adhesive properties.

Example 4

[0036] The method of Example 2 was repeated, using a glass substrate and 1H,1H,2H-perfluoro-1-octene (CF₃(CF₂)₅CH=CH₂) as the coating-forming material.

[0037] X-ray photoelectron spectroscopic analysis, FTIR analysis and contact angle measurements (with water and decane) were performed per Example 1 above, and results are shown in Table 5 below. The XPS and FTIR analysis showed that the glass substrate coating was rich in CF₂ and CF₃ and the contact angles for water and decane were determined as per example 1.

Table 5					
	XPS analysis			Contact Angle (water) (°)	Contact Angle (decane) (°)
	%C	%F	%O		
Theory	38.1	61.9	-	-	-
Example 4 coating	38.0	60.0	2.1	118.9 \pm 3.0	61.1 \pm 2.2

[0038] The contact angle measurement results show that the glass substrate has been rendered substantially hydrophobic and oleophobic by the coating.

CLAIMS

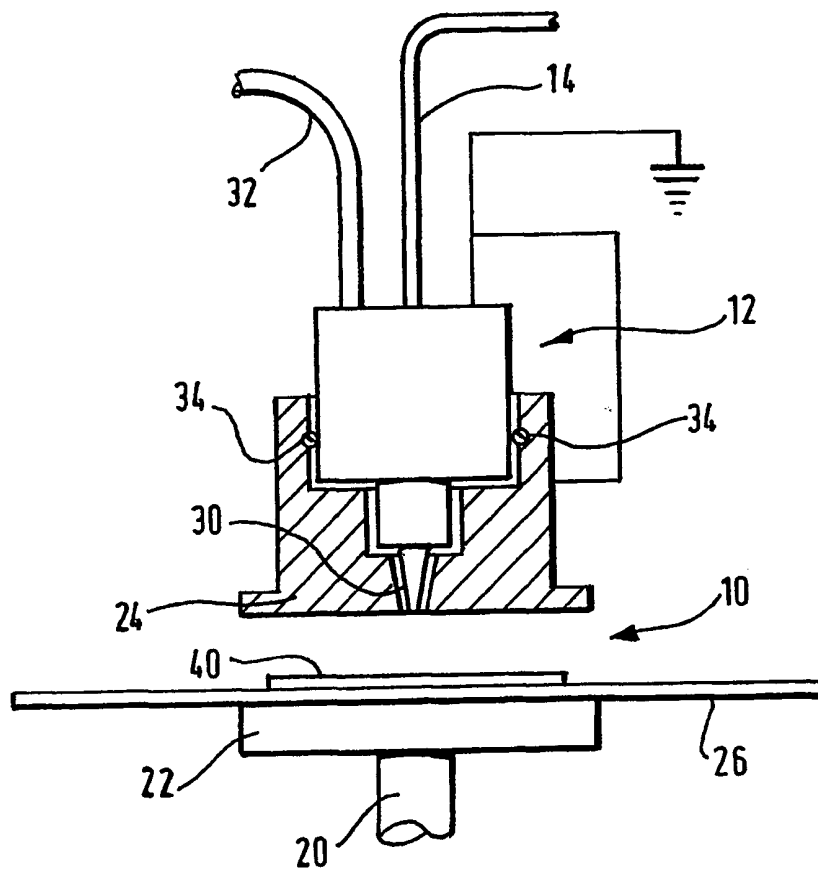
1. A method for forming a coating on a substrate, which method comprises introducing an atomised liquid and/or solid coating-forming material into an atmospheric pressure plasma discharge and/or an ionised gas stream resulting therefrom, and exposing the substrate to the atomised coating-forming material.
2. A method according to Claim 1 wherein the coating-forming material is introduced by direct injection.
3. A method according to Claim 1 or 2 wherein the coating-forming material is a silicon-containing material.
4. A method according to Claim 3 wherein the coating-forming material is selected from a dimethylsiloxane, and a siloxane having silicon-hydrogen bonds.
5. A method according to any preceding Claim wherein the plasma is generated in an oxygen containing atmosphere.
6. A method according to Claim 1 or 2 wherein the coating-forming material is an organic or organometallic material.
7. A method according to Claim 6 wherein the coating-forming material is selected from acrylic acid and a perfluoroalkene.

8. A method according to any preceding Claim wherein the substrate comprises metal, ceramic, plastics, woven or non-woven fibres, natural fibres, synthetic fibres, cellulosic material, and powder.
9. A method according to any preceding Claim wherein the coating increases the adhesive, release, gas barrier, moisture barrier, electrical and thermal conductivity, optical, dielectric, hydrophilic, hydrophobic, and/or oleophobic properties of the substrate.
10. A method of producing a substrate having a multi-layered coating according to any preceding claim whereby the coating is applied by repeatedly passing said substrate through the atmospheric plasma glow discharge or by passing said substrate through a series of atmospheric glow discharge chambers.
11. A method for polymerising a polymer forming material, which method comprises atomising the polymer-forming material, and exposing the atomised polymer-forming material to an atmospheric pressure plasma discharge.
12. A method according to any one of claims 1 to 4, 6 or 7 wherein the chemical properties of the atomised liquid and/or solid coating forming material are substantially retained in the resulting coating formed.
13. A method in accordance with any preceding claim wherein the substrate is coated continuously by use of a reel to reel apparatus.

14. A method in accordance with any preceding claim wherein the substrate is pre-treated by exposure to plasma prior to the introduction of coating forming material.
15. A method in accordance with any preceding claim wherein the coating formed on the substrate is post-treated by exposure to plasma.
16. A method in accordance with claim 14 or 15 wherein the plasma is applied by way of atmospheric pressure glow discharge.
17. A method in accordance with claim 16 wherein an oxygen containing material is added to the plasma.
18. A method in accordance with claim 17 wherein the oxygen containing materials are selected from the group of oxygen gas and water.
19. Apparatus for forming a coating on a substrate, which apparatus comprises means for generating an atmospheric pressure plasma glow discharge within which, in use, the substrate is placed, an atomiser for providing an atomised coating-forming material within the plasma discharge, and means for supplying a coating forming material to the atomiser.
20. Apparatus in accordance with claim 19 wherein the atomiser is an ultrasonic nozzle.

21. Apparatus in accordance with claim 19 or 20 wherein the substrate is fixed to a reel to reel apparatus to enable a continuous coating of the substrate.
22. A coated substrate prepared in accordance with the method of any one of claims 1 to 18.
23. A coated substrate in accordance with claim 22 wherein the chemical properties of the atomised liquid and/or solid coating forming material are retained in the resulting coating.
24. Use of a coated substrate formed in accordance with the method of any one of claims 1 to 18 as a lamination adhesive, an oxygen and/or moisture barrier or in organic light emitting diode devices.

FIG. 1



WO 02/028548 A3



SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW.

- (84) **Designated States (regional):** ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Declarations under Rule 4.17:

- *as to applicant's entitlement to apply for and be granted a patent (Rule 4.17(ii)) for the following designations AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL, PT, RO, RU, SD, SE,*

SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG)

- *of inventorship (Rule 4.17(iv)) for US only*

Published:

- *with international search report*

- (88) **Date of publication of the international search report:**

17 October 2002

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

INTERNATIONAL SEARCH REPORT

International Application No
PCT 01/04272

A. CLASSIFICATION OF SUBJECT MATTER
IPC 7 B05D7/24 B05D3/14

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 7 B05D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the International search (name of data base and, where practical, search terms used)
EPO-Internal, WPI Data, PAJ

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 5 290 378 A (KUSANO YUKIHIRO ET AL) 1 March 1994 (1994-03-01) column 3, line 14 - line 52 claims 1,5,6,12,13,17,18 ---	1,2,5,6, 8,9,11, 12, 17-19, 22-24
X	US 5 041 304 A (KUSANO YUKIHIRO ET AL) 20 August 1991 (1991-08-20) column 2, line 14 - line 31 column 2, line 61 - line 68 column 3, line 29 - column 4, line 38 claims --- -/-	1,2,6-9, 11,12, 19,22,23

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

* Special categories of cited documents:

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the international filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

T later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

X document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

Y document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

G document member of the same patent family

Date of the actual completion of the international search

27 June 2002

Date of mailing of the international search report

04/07/2002

Name and mailing address of the ISA
European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Stembrouck, I

INTERNATIONAL SEARCH REPORT

International Application No

PCT 01/04272

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 5 529 631 A (FUKUURA YUKIO ET AL) 25 June 1996 (1996-06-25) figure 1 column 1, line 51 - column 2, line 60 claims 1-4 ---	1,2,5,6, 8,9,11, 13, 17-19, 21,22
X	DE 195 46 187 A (FRAUNHOFER GES FORSCHUNG) 12 June 1997 (1997-06-12) column 1, line 61 - line 66 column 3, line 39 - line 45 figures claims 1,3 ---	1,6,8,9, 11,13, 19,21,22
X	US 4 957 062 A (SCHUURMANS HUBERTUS J A ET AL) 18 September 1990 (1990-09-18) column 1, line 44 - line 52 column 3, line 36 - line 43 column 5, line 8 - line 11 column 5, line 26 - line 29 claims ---	1-4,6,8, 9,11,19, 20,22
X	FR 2 713 511 A (AIR LIQUIDE) 16 June 1995 (1995-06-16) page 1, line 4 - line 12 page 1, line 34 - line 36 page 3, line 25 - line 31 page 4, line 12 - line 25 page 5, line 10 - line 21 claims 1,3,12 ---	1-4,6,8, 11,19,22
X	US 5 340 618 A (KOGOMA MASUHIRO ET AL) 23 August 1994 (1994-08-23) column 2, line 13 - line 48 column 4, line 20 - line 27 claim 1 ---	1,2,6,8, 9,11,19, 22,24
X	WO 99 08803 A (AGFA GEVAERT NV ; GATES ALLEN PETER (GB); BADYAL JAS PAL SINGH (GB)) 25 February 1999 (1999-02-25) page 2, line 1 - line 8 claim 1 ---	1,11,22

-/--

INTERNATIONAL SEARCH REPORT

International Application No

PCT 01/04272

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
P,X	WO 01 41942 A (CEBAL ;JUPIN ALAIN (FR); BENMALEK MOHAMED (FR)) 14 June 2001 (2001-06-14) page 5, line 7 -page 8, line 8 example 2 page 15, line 13 claims 1-4 ----	1-6,8,9, 11, 17-19, 22,24
E	WO 02 26401 A (CEBAL ;JUPIN ALAIN (FR); BENMALEK MOHAMED (FR)) 4 April 2002 (2002-04-04) page 5, line 5 - line 19 page 6, line 9 - line 15 page 7, line 27 - line 29 page 8, line 6 - line 21 claims 1,8 ----	1-6,8,9, 11, 17-19, 22,24
E	US 6 342 275 B1 (AKIYAMA HIROAKI ET AL) 29 January 2002 (2002-01-29) column 2, line 8 -column 4, line 15 column 7, line 32 - line 42 claims X & WO 95 18249 A (AKIYAMA HIROAKI) 6 July 1995 (1995-07-06) ----	1,2,5-8, 11,12, 17-19, 22,23
X		1,2,5-8, 11,12, 17-19, 22,23
E	WO 01 76773 A (KLAKE NIKLAS ;HOEPFNER KATRIN (DE); THYEN RUDOLF (DE); FRAUNHOFER) 18 October 2001 (2001-10-18) page 5, line 2 - line 10 page 6, line 19 - line 24 page 7, line 1 - line 2 page 8, line 19 -page 9, line 4 page 11, line 7 - line 10 claims 1,6 -----	1,6,8,9, 11,12, 22,23

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PC 01/04272

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
US 5290378	A	01-03-1994	JP 2897055 B2 JP 3263444 A	31-05-1999 22-11-1991
US 5041304	A	20-08-1991	JP 2990608 B2 JP 3202145 A	13-12-1999 03-09-1991
US 5529631	A	25-06-1996	JP 2811820 B2 JP 3143930 A	15-10-1998 19-06-1991
DE 19546187	A	12-06-1997	DE 19546187 A1	12-06-1997
US 4957062	A	18-09-1990	NONE	
FR 2713511	A	16-06-1995	FR 2713511 A1	16-06-1995
US 5340618	A	23-08-1994	JP 3283889 B2 JP 5023580 A	20-05-2002 02-02-1993
WO 9908803	A	25-02-1999	WO 9908803 A2 EP 1038049 A2 GB 2328692 A JP 2001515143 T US 6383575 B1	25-02-1999 27-09-2000 03-03-1999 18-09-2001 07-05-2002
WO 0141942	A	14-06-2001	FR 2801814 A1 AU 2523401 A WO 0141942 A2	08-06-2001 18-06-2001 14-06-2001
WO 0226401	A	04-04-2002	FR 2814382 A1 WO 0226401 A1	29-03-2002 04-04-2002
US 6342275	B1	29-01-2002	WO 9518249 A1	06-07-1995
WO 0176773	A	18-10-2001	DE 10017846 A1 WO 0176773 A2	25-10-2001 18-10-2001